

Atoms Family Atomic Math Challenge

Answer Key

Anything that has a mass—in other words, anything that occupies space—is composed of atoms. While its name originally referred to a particle that couldn't be divided any more—the smallest thing ...

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building block of all matter and chemistry. Atoms can combine with other atoms to form molecules but cannot be divided into smaller parts by ordinary chemical processes.

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In reality, they are composed of subatomic particles (protons, neutrons, and electrons).

Atoms are electrically neutral if they have an equal number of protons and electrons. Atoms that have either a deficit or a surplus of electrons are called ions. Electrons that are farthest from the nucleus ...

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to the concepts of atomic number and atomic mass. Nov 18, 2025 · Atoms are composed of three primary subatomic particles: protons, neutrons, and electrons. Protons and neutrons reside together in the dense, central region of the atom known as ...

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